Formation of Novel 1: 2 Adducts Between Bun₃P·CS₂ and Electron-deficient Alkynes

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The zwitterionic adduct between tri-n-butylphosphine and carbon disulphide reacts under neutral conditions with two equivalents of dimethyl and diethyl acetylenedicarboxylate, dibenzoylacetylene and methyl benzoylpropiolate to give novel crystalline dithiole-containing ylides whose structure is established by X-ray crystallography.

Although the red crystalline adducts formed between trialkylphosphines and CS_2 were first reported as early as 1860^1 and their zwitterionic phosphoniodithioformate structure was established by X-ray crystallography in 1961,² the cycloaddition reactions of this unusual type of 1,3-dipole have been little investigated. Reaction of 1 with acetylenic dipolarophiles such as dimethyl acetylenedicarboxylate (DMAD) does give 2-alkylidene-1,3-dithioles 3 if the addition is carried out in the presence of aromatic aldehydes to trap the initial 1 : 1 adducts 2 by a Wittig reaction.³ These intermediates can also be trapped by protonation, either intramolecularly⁴ to give 4 or by carrying out the addition in the presence of HBF₄ to give 5.⁵ Treatment of the stable salts 5 with Et₃N regenerates the phosphorane 2 thus allowing its use in the construction of a variety of tetrathiafulvalene derivatives of interest as donors for organic conductors.⁶ We now report that, in the absence of any trap, 1 reacts with a variety of electron-deficient alkynes in a 1:2 ratio to form stable crystalline adducts.

Reaction of 1 with DMAD in either ether or dichloromethane at room temperature for 18–24 h followed by



[†] For enquiries concerning the X-ray structure determination.



evaporation and column chromatography afforded an adduct as yellow prisms (32%, m.p. 114–115 °C, δ_P +21.9). Analytical and spectroscopic data clearly showed this to be a 1 : 2 adduct and the presence of four different ester groups initially led us to suspect that addition had occurred through the P and S atoms of 1 to give a structure like 6, 7 or 8. We particularly favoured 7 since this had good precedent in the intermediate 9 postulated by Huisgen⁷ in the reaction of Ph₃P and dimethyl azodicarboxylate with methyl propiolate. However none of these structures was in full accord with the ¹³C NMR spectroscopic data‡ and the real structure was revealed by an X-ray diffraction study§ to be 10.

In molecule **10** (Fig. 1), the atoms C(6) and C(7) are essentially coplanar with the dithiole ring plane [torsion angle S(1)-C(2)-C(6)-C(7) 1.1(6)°]. The relatively long P-C(7) [1.726(5) Å] and C(41)-O(41) [1.234(6) Å] distances together with the short value for C(7)-C(41) [1.402(7) Å] provide the first X-ray evidence for the contribution of the phosphonium enolate form **11** in an ester-stabilised phosphorane,¶ a feature consistent both with the only previous X-ray structures of

§ *Crystal data* for **10**: C₂₅H₃₉O₈PS₂. M = 562.69, crystal dimensions 0.28 × 0.36 × 0.51 mm, monoclinic space group $P2_1/n$, a = 13.039(3), b = 17.909(4), c = 13.226(2) Å β = 100.23(2)°, V = 3039.4 Å³, Z = 4, $D_c = 1.23$ g cm⁻³. μ (Mo-Kα radiation. $\lambda = 0.71073$ Å) = 2.6 cm⁻¹. Intensity data were measured at 21 °C using an Enraf-Nonius CAD4 diffractometer. Lorentz and polarisation corrections were applied to the data. The structure was solved by direct methods and refined by full-matrix least-squares calculations with hydrogens allowed for as riding atoms. At convergence R = 0.055, $R_w = 0.077$ for 2271 reflections with I > 3o(I). Atomic coordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre, See Notice to Authors, Issue No. 1.

¶ *Note added in proof:* after submission of our paper the structure of $Ph_3P=C(CO_2Et)CO(CH_2)_3CO_2H$ was reported, but in that case the delocalisation is mainly into the keto carbonyl group; A. D. Abell, J. Trent and W. T. Robinson, *J. Chem. Soc., Chem. Commun.*, 1991, 362.



Fig. 1 An ORTEP projection of 10. [C(41), the atom bonded to C(7). O(41) and O(42), is not labelled for space considerations]. Selected bond lengths (Å) and angles (°): P-C(7) 1.726(5); C(7)–C(41) 1.402(7); C(41)–O(41) 1.234(6); C(41)–O(42) 1.350(6); C(7)–C(6) 1.492(6); C(6)–C(31) 1.450(7); C(31)–O(31) 1.218(6); C(6)–C(2) 1.360(7); C(2)–S(1) 1.733(5); S(1)–C(5) 1.742(5); C(4)–C(5) 1.331(7); P-C(7)-C(41) 116.8(3); C(7)–C(41)–O(41) 126.2(5); P-C(7)-C(6) 120.4(4); C(7)–C(6)–C(31) 123.9(4); C(7)–C(6)–C(2) 118.8(4); S(1)–C(2)–S(3) 114.5(3); C(2)–S(1)–C(5) 95.2(3); S(1)–C(5)–C(4) 117.4(4); torsion angle P-C(7)-C(41)-O(41) 0.0(0.7).



 β -oxophosphoranes⁸ and with reactivity, dipole moment, UV⁹ and IR studies.¹⁰ The compound showed typical reactivity for a stabilised yield: it could be readily converted by anhydrous acids to phosphonium salts **12** which reformed **10** upon treatment with triethylamine. It was resistant to Wittig reaction with 2,4-dinitrobenzaldehyde under forcing conditions.

The formation of this product is readily explained by initial reaction to give the 1:1 adduct of type **2** which then undergoes [2 + 2] cycloaddition with a second DMAD molecule to give phosphacyclobutene **13** followed by electrocyclic ring-opening to give **10**. DMAD is already known to react in this way with phosphinimines, Ph₃P=NAr,¹¹ and with ylides such as Ph₃P=CHPh and Ph₃P=CPh₂¹² while both stabilised ylides and phosphinimines are reported to undergo similar [2 + 2] cycloaddition followed by ring-opening with activated nitriles.¹³ It is notable that only the 1:2 adduct **10** was formed even in the presence of excess of **1** implying that the initial 1:1 adduct **2** reacts much more readily with DMAD than does **1**.

Reaction of 1 with other electron-deficient alkynes gave adducts of similar structure: diethyl acetylenedicarboxylate gave 14 (43%, m.p. 90–91 °C, δ_P +21.4) and dibenzoylacetylene gave 15 (12%, m.p. 161–162 °C, δ_P +19.5). In the case of the unsymmetrical alkyne PhCOC=CCO₂Me only two of the four possible isomers were formed in a 3:2 ratio (23%, m.p. 95–96 °C, δ_P +20.5, 20.4). The presence of a two-bond phosphorus coupling (≈8.5 Hz) on one PhCO signal of each isomer showed these to be the *E*- and *Z*-isomers 16 pointing to complete regiospecificity in the [2 + 2] cycloaddition step. Further results on the cycloaddition of 1 with other types of dipolarophile will be reported shortly.

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 $[\]frac{1}{2} \delta_{\rm C}$ (75 MHz, CDCl₃) 13.6, 22.1[55], 24.2[15], 24.6[4], 42.1[122], 50.0, 51.5, 53.0, 53.1, 109.3[9], 130.7, 136.2, 160.4, 160.45, 161.7[4], 167.8 and 169.2[20] (Figures in square brackets are $J_{\rm CP}$ in Hz).

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References

- 1 A. W. Hofmann, Liebigs Ann. Chem. Suppl., 1860, 1, 1.
- 2 T. N. Margulis and D. H. Templeton, J. Am. Chem. Soc., 1961, 83, 995.
- 3 H. D. Hartzler, J. Am. Chem. Soc., 1971, 93, 4961.
- 4 C. U. Pittman and M. Narita, J. Chem. Soc., Chem. Commun., 1975, 960.
- 5 M. Sato, N. C. Gonella and M. P. Cava, J. Org. Chem., 1979, 44, 930.

- 6 T. Sugimoto, H. Awaji, I. Sugimoto, Y. Misaki, T. Kawase, S. Yoneda, Z. Yoshida, T. Kobayashi and H. Anzai, *Chem. Mater.*, 1989, 1, 535; Y. Misaki, Y. Matsumura, T. Sugimoto and Z. Yoshida, *Tetrahedron Lett.*, 1989, 30, 5289.
- 7 E. Brunn and R. Husigen, Angew. Chem., Int. Ed. Engl., 1969. 8, 513.
- 8 A. J. Speziale and K. W. Ratts, J. Am. Chem. Soc., 1965, 87, 5603; F. S. Stephens, J. Chem. Soc., 1965, 5640, 5658.
- 9 F. Ramirez and S. Dershowitz, J. Org. Chem., 1957, 22, 41.
- 10 A. J. Speziale and K. W. Ratts, J. Org. Chem., 1963, 28, 465; J. Am. Chem. Soc., 1963, 85, 2790.
- 11 G. W. Brown, J. Chem. Soc. (C), 1967, 2018.
- 12 H. J. Bestmann and O. Rothe, Angew. Chem., Int. Ed. Engl., 1964, 3, 512; J. B. Hendrickson, C. Hall, R. Rees and J. F. Templeton, J. Org. Chem., 1965, 30, 3312.
- 13 E. Ciganek, J. Org. Chem., 1970, 35, 3631; H. Trabelsi, E. Bollens and A. Cambon, Synthesis, 1990, 623.